

**REMARKS**

In the Office Action dated October 15, 2004, the Examiner rejected all pending claims (1, 2, 6, and 7) as obvious over U.S. Patent No. 4,182,437 ("Roberts et al.") in view of EP 1081406A2 ("Yamane"). The Examiner treated Yamane as a prior art reference under 35 U.S.C. § 102(a), and stated that "[w]ith regards to the foreign priority date claimed in the present application, it is noted that no certified English translation is in the file." (Detailed Action at 2.)

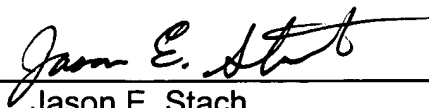
Applicants submit the attached certified English translation of Japanese Patent Application 2000-211567. This document serves to establish Applicants' right to a priority date of July 12, 2000, and thus removes Yamane (which published on March 7, 2001) as a prior art reference under 35 U.S.C. § 102(a). Because the Examiner's obviousness rejection is based in part on a reference that is not prior art to the present application, the Examiner has not presented a *prima facie* case of obviousness. Withdrawal of this rejection and timely allowance of the pending claims is respectfully requested.

Please grant any extensions of time required to enter this response and charge any additional required fees to our deposit account 06-0916.

Respectfully submitted,

FINNEGAN, HENDERSON, FARABOW,  
GARRETT & DUNNER, L.L.P.

Dated: November 15, 2004

By:   
Jason E. Stach  
Reg. No. 54,464

**Enclosure:** Certified English Translation of  
Japanese Patent Application 2000-211567

**STATEMENT**

I, Makoto KONDO, of c/o NGB Corporation, ARK Mori Building 13F, 12-32, Akasaka 1-Chome, Minato-ku, Tokyo 107-6013 Japan, hereby state that I am conversant with both the English and Japanese languages and certify to best of my knowledge and belief that the attached is a true and correct English translation of the priority document of Japanese patent application 2000-211567 filed on July 12, 2000.

Date: November 11, 2004

近藤 誠

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Makoto KONDO

JAPAN PATENT OFFICE

This is to certify that the annexed is a true copy of the following application as filed with this Office.

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Application Number: Patent Application No. 2000-211567

Applicant(s): AKEBONO BRAKE INDUSTRY CO., LTD.

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Commissioner,  
Japan Patent Office      Kouzo OIKAWA (Seal)

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[Inventor]

[Address or Residence] c/o AKEBONO BRAKE INDUSTRY CO., LTD.,  
19-5, Nihonbashi Koami-cho, Chuo-ku,  
Tokyo

[Name] Akihiro HIKICHI

[Inventor]

[Address or Residence] c/o AKEBONO BRAKE INDUSTRY CO., LTD.,  
19-5, Nihonbashi Koami-cho, Chuo-ku,  
Tokyo

[Name] Mikiya HARUTA

[Applicant for Patent]

[Indication Number] 000000516

[Name or Appellation] AKEBONO BRAKE INDUSTRY CO., LTD.

[Agent]

[Indication Number] 100105647

[Patent Attorney]

[Name or Appellation] Shohei OGURI

[Telephone Number] 03-5561-3990

[Selected Agent]

[Indication Number] 100105474

[Patent Attorney]

[Name or Appellation] Hironori HONDA

[Telephone Number] 03-5561-3990

[Selected Agent]

[Indication Number] 100108589  
[Patent Attorney]  
[Name or Appellation] Toshimitsu ICHIKAWA  
[Telephone Number] 03-5561-3990  
[Selected Agent]  
[Indication Number] 100115107  
[Patent Attorney]  
[Name or Appellation] Takeshi TAKAMATSU  
[Telephone Number] 03-5561-3990  
[Selected Agent]  
[Indication Number] 100090343  
[Patent Attorney]  
[Name or Appellation] Yuriko KURIU  
[Telephone Number] 03-5561-3990  
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[Name of Document] SPECIFICATION

[Title of the Invention] NON-ASBESTOS FRICTION MATERIAL

[Claims for Patent]

[Claim 1] A non-asbestos friction material  
5 comprising a fibrous reinforcement, a friction modifier, and  
a binder, characterized in that  $\text{SiO}_2$  as a principal component,  
and a soluble amorphous substance having a composition composed  
of 18 to 40 wt% of at least one kind of  $\text{CaO}$  and  $\text{MgO}$ , less than  
10 wt% of at least one kind of  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$ , and less than  
2 wt% of at least one kind of  $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ ,  $\text{FeO}$  and  $\text{Fe}_2\text{O}_3$  are mixed  
10 as friction material components.

[Claim 2] The non-asbestos friction material  
according to Claim 1, characterized in that said soluble  
amorphous substance is mixed in a range of from 1 wt% to 30  
15 wt% of a total of said friction material.

[Claim 3] The non-asbestos friction material  
according to Claim 1, characterized in that said soluble  
amorphous substance is fibrous or granular.

[Claim 4] The non-asbestos friction material  
20 according to Claim 1, characterized in that said soluble  
amorphous substance is formed of fibers having an average fiber  
diameter in a range of from 2  $\mu\text{m}$  to 9  $\mu\text{m}$  and an average fiber  
length in a range of from 100  $\mu\text{m}$  to 1,500  $\mu\text{m}$ .

[Claim 5] The non-asbestos friction material  
25 according to Claim 1, characterized in that said soluble  
amorphous substance is formed of grains having an average grain

size in a range of from 2  $\mu\text{m}$  to 100  $\mu\text{m}$ .

[Detailed Description of the Invention]

[0001]

[Technical Field Pertinent to the Invention]

5           The present invention relates to a friction material,  
and particularly relates to a friction material for an industrial  
machine, a railway vehicle, a baggage vehicle, a passenger  
vehicle, or the like. More particularly, the present invention  
relates to a brake pad, a brake lining, a clutch facing, etc.,  
10   for the aforementioned applications.

[0002]

[Conventional Art]

          In a friction material principally for a brake or the  
like, a fibrous reinforcement is used as one of materials for  
15   enhancing the strength of the friction material. As such a  
fibrous reinforcement, ceramic fibers (rock wool, slag wool),  
glass fibers, steel fibers, aramid fibers, potassium titanate  
fibers, etc. are available. Since these fibers have their  
own properties, several kinds of these fibers are mixed in  
20   use.

          Of these fibers, rock wool is rated highly as an abrasive  
hard inorganic fibrous reinforcement. That is, rock wool  
enhances the strength and the heat resistance of the friction  
material as a whole and improves the wear resistance. At the  
25   same time, it heightens the friction coefficient of the friction  
material due to its abrasive property. Particularly, rock



wool can ensure a high friction coefficient at the time of a high load, for example, at the time of high-speed braking.

[0003]

[Problem to be Solved by the Invention]

5           Rockwool and slag wool are fibrous amorphous substances. Now, according to Q1 of EU Commission 97/69/EC, soluble amorphous fibers harmless to human bodies are defined to be mineral fibers having a length longer than 20  $\mu\text{m}$  and having a weighted half life shorter than 40 days on the basis of a  
10   short-term in vivo retentivity test by endotracheal injection.

[0004]

          However, the chemical composition (wt%) of rock wool is typically of 35 to 45 of  $\text{SiO}_2$ , 10 to 20 of  $\text{Al}_2\text{O}_3$ , 30 to 40 of  $\text{CaO}$ , 4 to 8 of  $\text{MgO}$ , 1 to 4 of  $\text{MnO}$ , and 0.1 to 3 of  $\text{Fe}_2\text{O}_3$ .

15   Thus, in spite of being an amorphous substance, rock wool has the content of  $\text{Al}_2\text{O}_3$  in a range of from 10 wt% to 20 wt% to be so high that the rock wool is difficult to be soluble in vivo. As a result, it does not come under the above-mentioned definition.

20           [0005]

          It is an object of the present invention to obtain a non-asbestos friction material without using any ceramic fiber such as rock wool, or the like, which is undesirable on the working environmental sanitation. Nevertheless the  
25   non-asbestos friction material provides friction properties (effectiveness adjustment, rust removability, and so on) and

strength equivalent to those in the case where such ceramic fibers having a high content of the alumina component, and at the same time, it can also restrain the manufacturing cost from increasing.

5 [0006]

[Means for Solving Problem]

To solve the foregoing problems, the present inventors carried out various researches on materials or forms of inorganic substances which use no ceramic fiber such as rock wool or  
10 the like undesirable on the working environmental sanitation, but which nevertheless provide a friction material with friction properties and strength equivalent to those in the case where such ceramic fibers are used. The problem in the working environmental sanitation is caused by rock wool or slag wool  
15 because the composition thereof has a high content of the alumina component in a range of from 10 wt% to 20 wt%.

The present inventors therefore attained the present invention in the following point of view. That is, even if an amorphous inorganic substance containing only a small amount  
20 of alumina is used, there still can be friction properties and strength equivalent to those in the case where the above-mentioned amorphous inorganic substance such as rock wool or the like is used. In addition, the solubility of the amorphous inorganic substance in humor is enhanced so that  
25 health anxiety can be avoided.

[0007]

That is, the present invention solved the foregoing problems by the following means.

(1) A non-asbestos friction material including a fibrous reinforcement, a friction modifier, and a binder, characterized in that  $\text{SiO}_2$  as a principal component, and a soluble amorphous substance having a composition composed of 18 to 40 wt% of at least one kind of  $\text{CaO}$  and  $\text{MgO}$ , less than 10 wt% of at least one kind of  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$ , and less than 2 wt% of at least one kind of  $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ ,  $\text{FeO}$  and  $\text{Fe}_2\text{O}_3$  are mixed as friction material components.

(2) A non-asbestos friction material according to the paragraph (1), characterized in that the soluble amorphous substance is mixed in a range of from 1 wt% to 30 wt% of a total of the friction material.

(3) A non-asbestos friction material according to the paragraph (1), characterized in that the soluble amorphous substance is fibrous or granular.

(4) A non-asbestos friction material according to the paragraph (1), characterized in that the soluble amorphous substance is formed of fibers having an average fiber diameter in a range of from 2  $\mu\text{m}$  to 9  $\mu\text{m}$  and an average fiber length in a range of from 100  $\mu\text{m}$  to 1,500  $\mu\text{m}$ .

(5) A non-asbestos friction material according to the paragraph (1), characterized in that the soluble amorphous substance is formed of grains having an average grain size in a range of from 2  $\mu\text{m}$  to 100  $\mu\text{m}$ .

[0008]

[Mode for Carrying Out the Invention]

A friction material is constituted by a fibrous reinforcement, a friction modifier, a binder, and so on. According to the present invention, not only asbestos but also in vivo insoluble amorphous substances which are amorphous inorganic substances such as rock wool, slag wool, or the like, undesirable on the working environmental sanitation, and which contains 10 to 20 wt% of alumina, are not used as fibrous reinforcements. In place of such fibrous reinforcements, a fibrous or granular soluble amorphous substance containing not higher than 5 wt% of alumina is used. Incidentally, the word "soluble" means that the substance is soluble in humor when it enters a body.

15

[0009]

The soluble amorphous substance according to the present invention is composed of  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{ZrO}_2$ , oxides of Na, K, Ca, Mg and Ba, and so on. Preferably, the total amount of the oxides of Na, K, Ca, Mg and Ba exceeds 18wt%.

20

In the soluble amorphous substance composed of such chemical components, as the component ratio of  $\text{Al}_2\text{O}_3$  or  $\text{ZrO}_2$  is lower, and as the ratio of CaO or MgO is higher, the in-vivo solubility is superior.

That is, in the range of  $\text{Al}_2\text{O}_3 \leq 5\text{wt}\%$  and  $\text{ZrO}_2 \leq 5\text{wt}\%$ , the solubility is more improved as they approach 0 wt%.

25

On the other hand, as the sum of CaO and MgO becomes

large, the solubility is more improved.

[0010]

However, if the ratio of CaO and MgO is made too high, the heat resistance is lowered. It is therefore desired that  
5 the total ratio of CaO and MgO is not higher than 40wt%.

In addition, as trace components, the following oxides, that is, Na<sub>2</sub>O, K<sub>2</sub>O, FeO and Fe<sub>2</sub>O<sub>3</sub> may be contained by less than 2 wt% as the total amount of them.

An example of a preferred soluble amorphous substance  
10 is a SiO<sub>2</sub>-ZrO<sub>2</sub>-CaO-MgO amorphous inorganic substance.

Examples of typical compositions (wt%) include 64.5 of SiO<sub>2</sub>, 5.0 of ZrO<sub>2</sub>, 17.0 of CaO, and 13.5 of MgO; 65.0 of SiO<sub>2</sub>, 29.5 of CaO, and 5.5 of MgO; 65.0 of SiO<sub>2</sub>, 19.5 of CaO, and 15.5 of MgO; 65 of SiO<sub>2</sub>, 0.3 of Al<sub>2</sub>O<sub>3</sub>, 31.1 of CaO, 3.2  
15 of MgO, and 0.3 of Fe<sub>2</sub>O<sub>3</sub>; and so on.

The proportion of the soluble amorphous substance as a friction material component is set to be in a range of from 1 wt% to 30 wt%. If the loading is smaller than 1 wt%, it is difficult to disperse the soluble amorphous substance  
20 uniformly when it is mixed and stirred, so that a desired effect cannot be obtained. On the contrary, if the loading is larger than 30 wt%, the effect is improved, but a partner material is worn harder. Thus, the loading is set to be not larger than 30 wt%. Preferably, the loading is set to be in a range  
25 of from 2.5 wt% to 30 wt%.

[0011]

Fibers of the soluble amorphous substance according to the present invention have an average fiber diameter in a range of from 2  $\mu\text{m}$  to 9  $\mu\text{m}$ , preferably in a range of from 3  $\mu\text{m}$  to 6  $\mu\text{m}$ . If the average fiber diameter is not larger than 2  $\mu\text{m}$ , it becomes difficult to manufacture the fibers so that they are not economical. If the average fiber diameter is not smaller than 9  $\mu\text{m}$ , the dispersibility of the fibers when they are mixed with other materials deteriorates. In addition, when the fibers are made into a friction material, the aggressiveness of the friction material against a partner metal to be rubbed deteriorates. In addition, the average fiber length is in a range of from 100  $\mu\text{m}$  to 1,500  $\mu\text{m}$ , preferably in a range of from 500  $\mu\text{m}$  to 1,000  $\mu\text{m}$ . If the average fiber length is shorter than 100  $\mu\text{m}$ , the reinforcement effect cannot be obtained sufficiently. If the average fiber length is longer than 1,500  $\mu\text{m}$ , the dispersibility of the fibers when they are mixed with other materials deteriorates.

[0012]

Grains of the soluble amorphous substance according to the present invention have an average grain size in a range of from 2  $\mu\text{m}$  to 100  $\mu\text{m}$ , preferably in a range of from 5  $\mu\text{m}$  to 30  $\mu\text{m}$ . If the average grain size is not larger than 2  $\mu\text{m}$ , it becomes difficult to manufacture the grains so that they are not economical. If the average grain size is not smaller than 30  $\mu\text{m}$ , the aggressiveness of a friction material made from the grains against a partner metal to be rubbed deteriorates.

[0013]

In the friction material according to the present invention, examples of fibrous reinforcements include organic fibers such as aromatic polyamide fibers, fire-resistant acrylic fibers, or the like; metal fibers such as copper fibers, steel fibers, or the like; and inorganic fibers such as potassium titanate fibers,  $\text{Al}_2\text{O}_3\text{-SiO}_2$  ceramic fibers, or the like.

Examples of inorganic fillers include metal particles of copper, aluminum, zinc, or the like; scaly inorganic substances such as vermiculite, mica, or the like; barium sulfate; calcium carbonate; etc.

Examples of thermosetting resin binders include phenolic resin (including straight phenolic resin, and variously modified phenolic resins affected by rubber or the like); melamine resin; epoxy resin; polyimide resin; etc.

In addition, examples of friction modifiers include inorganic friction modifiers such as alumina, silica, magnesia, zirconia, chrome oxide, quartz, or the like; and organic friction modifiers such as synthetic rubber, cashew resin, or the like.

Examples of solid lubricants include graphite, molybdenum disulfide, etc.

Various composition ratios can be adopted as the composition of the friction material.

That is, these friction materials may be used individually or in combination of two or more kinds in accordance with friction properties required of a product, for example,

a friction coefficient, wear resistance, a vibration property, a friction noise property, and so on.

[0014]

In a process for manufacturing a brake pad for a disc  
5 brake, a pressure plate is molded into a predetermined shape  
by sheet metal pressing, subjected to degreasing processing  
and primer processing, and coated with an adhesive agent.  
A fibrous reinforcement of heat-resistant organic fibers, metal  
fibers or the like and a powdered raw material of an  
10 organic/inorganic filler, a friction modifier, a thermosetting  
resin binder, and so on are mixed with each other and sufficiently  
homogenized by stirring. The mixture of the fibrous  
reinforcement and the powdered raw material are formed  
(preformed) at room temperature and at predetermined pressure  
15 so that a preformed friction material is produced. The pressure  
plate and the preformed friction material are thermoformed  
at predetermined temperature and pressure in a thermoforming  
process so that both the members are fixed integrally with  
each other. The integrated friction material is after-cured  
20 and finally subjected to finishing processing. The  
manufacturing steps to this point are the same as those in  
the background-art method.

[0015]

[Examples]

25 The present invention will be described specifically  
on the basis of its examples. However, the present invention





an average grain size of 10  $\mu$ m was used.

[0017]

[Experiments on the soluble amorphous fibers]

(Compositions of the friction material samples)

5           Compositions of friction materials having mixture ratios shown in the following Table 1 were used for producing samples of the friction materials. Thus, Samples of Examples 1 to 3 were produced.

10           Incidentally, in order to make comparison, samples containing no soluble amorphous fiber (Comparative Examples 1 and 2) were also produced.

(Producing Samples of brake pads)

15           Brake pads of the friction material samples having the above-mentioned compositions were produced by a background-art producing method.

[0018]

[Table 1]

Table 1                      (mixture: wt%)

Friction Material Component	Comparative Examples		Examples		
	1	2	1	2	3
Binder resin	10	10	10	10	10
Friction dust	10	10	10	10	10
Barium sulfate	45	50	45	45	45
Zirconia	2	2	2	2	2
Graphite	8	8	8	8	8

Copper fibers / Aramid fibers	10/5	10/5	10/5	10/5	10/5
Potassium titanate fibers	5	5	5	5	5
Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> ceramic fibers	5	-	-	2.5	-
Soluble amorphous fibers	-	-	5	2.5	30

[0019]

(Examination method)

With a full-size dynamometer as a test machine, the following examinations were performed upon a disc brake using a rotor. Incidentally, as a brake pad, a test piece was used.

(1) JASO effect evaluation

Friction coefficient and JASO effect evaluation were made on the conditions of an initial speed of 50 km/h, an initial speed of 100 km/h, an initial speed of 130 km/h, and a deceleration  $\alpha=5.88 \text{ m/s}^2$ .

(2) JASO first-fade minimum  $\mu$

(3) Rust removing ratio

A rotor with about 50 $\mu\text{m}$ -thick rust was rubbed with a friction material, and the rust removing ratio was measured after the performance of N(number of braking applications) =200 (times).

If the rust removing ratio is not lower than 80%, the rust removing is accepted.

(4) T/P braking rotor aggressiveness

Rotor wear quantity ( $\mu\text{m}$ ) per 1,000 times on the conditions of an initial speed of 50 km/h, a deceleration

$\alpha=0.98\text{m/s}^2$  and a temperature of  $100^\circ\text{C}$ .

(Examination results)

The results of the examinations are shown in Table

2.

5

[0020]

[Table 2]

Table 2

		Comparative Examples		Examples		
		1	2	1	2	3
JASO Effect	50km/h	0.45	0.38	0.44	0.45	0.46
	100km/h	0.40	0.30	0.40	0.41	0.41
	130km/h	0.33	0.23	0.36	0.36	0.37
JASO 1st Fade minimum $\mu$		0.24	0.20	0.26	0.27	0.28
Rust removing ratio (%)		100	50	100	100	100
Rotor aggressiveness ( $\mu\text{m}$ )		5.5	2.0	3.2	3.7	4.6

[0021]

Examples 4 to 7 and Comparative Examples 3 and 4

10 [Experiments on the soluble amorphous powder]

(Compositions of the friction material samples)

Compositions of friction materials having mixture ratios shown in the following Table 3 were used for producing samples of friction materials. Thus, Samples of Examples 4

15 to 7 were produced.

In order to make comparison, samples containing no soluble amorphous powder (Comparative Examples 3 and 4) were also produced.

(Producing samples of brake pads)

- 5 Brake pads of the friction material samples having the above-mentioned compositions were produced by a background-art producing method.

[0022]

[Table 3]

10

Table 3

(mixture: wt%)

	Comparative Examples		Examples			
	3	4	4	5	6	7
Binder resin	10	10	10	10	10	10
Friction dust	10	10	10	10	10	10
Barium sulfate	45	50	45	45	20	45
Zirconia	3	3	3	3	3	0
Graphite	7	7	7	7	7	7
Copper fibers / Aramid fibers	10/5	10/5	10/5	10/5	10/5	10/5
Potassium titanate fibers	5	5	5	5	5	5
Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> ceramic fibers	5	-	-	2.5	-	5
Soluble amorphous powder	-	-	5	2.5	30	3

[0023]

(Examination method)

In the same manner as in the examples where the fibers were mixed, the following examinations were performed upon a disc brake using a rotor.

(1) JASO effect evaluation

5 Friction coefficient and JASO effect evaluation were made on the conditions of an initial speed of 50 km/h, an initial speed of 100 km/h, an initial speed of 130 km/h, and a deceleration  $\alpha=0.6G$ .

10 (2) JASO first-fade minimum  $\mu$  (the same as that in the case of the fibers)

(3) Rust removing ratio (the same as that in the case of the fibers)

(4) T/P braking rotor aggressiveness (the same as that in the case of the fibers)

15 (Examination results)

The results of the examinations are shown in Table 4.

[0024]

[Table 4]

20

Table 4

		Comparative Examples		Examples			
		3	4	4	5	6	7
JASO Effect	50km/h	0.45	0.38	0.42	0.42	0.45	0.43
	100km/h	0.41	0.30	0.40	0.41	0.43	0.41
	130km/h	0.35	0.24	0.36	0.36	0.38	0.35

JASO 1st Fade minimum $\mu$	0.25	0.20	0.27	0.27	0.28	0.26
rust removing ratio (%)	100	50	100	100	100	100
T/P Rotor aggressiveness ( $\mu\text{m}$ )	5.9	2.1	3.6	3.9	4.8	4.3

[0025]

[Effects of the Invention]

According to the present invention, by using fibers or powder of a soluble amorphous inorganic substance in which the content of alumina ( $\text{Al}_2\text{O}_3$ ) and zirconia ( $\text{ZrO}_2$ ) lowering the solubility in vivo is lower than 10 wt%, preferably not higher than 5 wt%, it is possible to manufacture a friction material having the following effects. That is, the rust removing performance equivalent to that of rock wool or slag wool is ensured. At the same time, the rotor aggressiveness is reduced, and the effect stability is improved. Further, the solubility in vivo is so high that the environmental safety can be improved. Thus, it is possible to avoid a problem caused by use of ceramic fibers such as rock wool, slag wool, or the like.

[Name of Document] ABSTRACT

[Abstract]

[Problem] To obtain a non-asbestos friction material without using any ceramic fiber such as rock wool, or the like, which has a high content of an alumina component and which is undesirable on the working environmental sanitation. Nevertheless the non-asbestos friction material provides friction properties and strength equivalent to those in the case where such ceramic fibers are used, and can restrain the manufacturing cost from increasing.

[Solving Means] A non-asbestos friction material including a fibrous reinforcement, a friction modifier, and a binder, characterized in that  $\text{SiO}_2$  as a principal component, and a soluble amorphous substance having a composition composed of 18 to 40 wt% of at least one kind of  $\text{CaO}$  and  $\text{MgO}$ , less than 10 wt% of at least one kind of  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$ , and less than 2 wt% of at least one kind of  $\text{Na}_2\text{O}$ ,  $\text{K}_2\text{O}$ ,  $\text{FeO}$  and  $\text{Fe}_2\text{O}_3$  are mixed as friction material components. Preferably, the soluble amorphous substance is mixed in a range of from 1 wt% to 30 wt% of a total of the friction material. Preferably, the soluble amorphous substance is fibrous or granular.

[Selected Drawing] None